Colopsinols B and C, new long chain polyhydroxy compounds from cultured marine dinoflagellate *Amphidinium* sp.

Takaaki Kubota, "Masashi Tsuda," Miho Takahashi, "Masami Ishibashi," Hideo Naoki^b and Jun'ichi Kobayashi *"

^{*a*} Graduate School of Pharmaceutical Sciences, Hokkaido University, Sapporo 060-0812, Japan ^{*b*} Suntory Institute for Bioorganic Research, Shimamoto-cho, Mishima-gun, Osaka 618-8503

Received (in Cambridge, UK) 3rd August 1999, Accepted 29th September 1999

Colopsinols B (1) and C (2), unique polyhydroxy compounds consisting of a C_{53} -linear aliphatic chain with three C_1 branches, three ether rings, six hydroxy groups, a glucoside moiety, and a sulfate ester have been isolated from the cultured marine dinoflagellate *Amphidinium* sp. The structures of 1 and 2 were elucidated on the basis of extensive NMR techniques as well as FABMS/MS data of the peroxide (3) of 2.

During our search for bioactive metabolites from marine dinoflagellates belonging to the genus *Amphidinium*,¹ we have isolated colopsinol A, a new class of polyhydroxy compound possessing inhibitory activity against DNA polymerase α and β .² Further investigation of the extract of this strain resulted in the isolation of two new polyhydroxy compounds, colopsinols B (1) and C (2), possessing a C₅₃-linear aliphatic chain with three C₁ branches, three ether rings, six hydroxy groups, a glucoside moiety, and a sulfate ester. This paper describes the isolation and structure elucidation of 1 and 2, which possess a different long aliphatic chain from that of colopsinol A.

The harvested cells of the dinoflagellate \dagger were extracted with MeOH-toluene (3:1). The toluene-soluble materials of the extract were subjected to a silica gel column (CHCl₃-MeOH, 95:5 \rightarrow MeOH). The fraction eluting with MeOH was separated by gel filtration on Sephadex LH-20 (CHCl₃-MeOH, 1:1) and centrifugal partition chromatography (CPC, ascended

[†] The dinoflagellate *Amphidinium* sp. (strain number Y-5) is a symbiont of an Okinawan marine acoel flatworm *Amphiscolops* sp., from which amphidinolides A–D, J, K, and M–S and amphidinin A were previously isolated.¹



Stereochemistry of the tetrahydropyran and the tetrahydrofuran rings is relative, and the epoxide ring is trans.

J. Chem. Soc., *Perkin Trans.* 1, 1999, 3483–3487 3483

This journal is © The Royal Society of Chemistry 1999



mode, CHCl₃–MeOH–H₂O, 5:6:4) followed by C₁₈ column chromatography (*i*-PrOH–H₂O) to yield colopsinol B (1, 0.001%, wet weight), together with a known related compound, colopsinol A.² Another fraction of CPC was separated by C₁₈ column chromatography to obtain colopsinol C (2, 0.0002%) and the peroxide form (3, 0.0008%) of 2, the latter of which seemed to have been derived from colopsinol C (2) through autooxidation of the s-*cis* diene part of 2 during purification or presevation.

Colopsinol B (1) has the molecular formula of $C_{68}H_{113}$ -O25SNa as revealed by the positive ion HRESIMS [m/z 1407.7040, $(M+Na)^+$, $\Delta -4.7$ mmu]. The UV absorption at 232 nm (ε 13000 dm³ mol⁻¹ cm⁻¹) was attributed to a diene chromophore, and IR absorptions were indicative of the presence of hydroxy (v_{max} 3430 cm⁻¹), ketone (v_{max} 1700 cm⁻¹), and sulfate (v_{max} 1250 cm⁻¹) functionalities. FABMS fragment ions at $m/z \ 80 \ (SO_3^-)$ and 97 (HSO₄⁻) suggested that 1 had a sulfate ester. ¹H and ¹³C NMR data (Table 1) disclosed that 1 contained a ketone, two sp² quaternary carbons, seven sp² methines, three sp² methylenes, twenty-four oxymethines, two oxymethylenes, one sp³ methine, twenty-six sp³ methylenes, and two methyl groups. A ¹³C NMR deuterium-induced shift experiment using CD₃OH and CD₃OD revealed the presence of thirteen hydroxy groups, seven of which were due to two sugar moieties. Of thirteen unchanged oxygenated carbons two lowfield ones (C-1', $\delta_{\rm C}$ 105.03; C-1", $\delta_{\rm C}$ 105.86 in CD₃OD) were assigned as anomeric carbons of sugar units. The presence of an epoxide ring was implied by two high-field carbons at $\delta_{\rm C}$ 61.31 (C-22) and $\delta_{\rm C}$ 61.62 (C-23). Four oxymethines (C-4, $\delta_{\rm C}$ 73.77; C-8, $\delta_{\rm C}$ 74.91; C-45, $\delta_{\rm C}$ 88.46; C-48, $\delta_{\rm C}$ 79.03) of the unchanged carbons were elucidated to constitute two ether linkages from 2D NMR data as described below. The three unchanged oxygenated carbons were indicated to be those bearing glycoside linkages (C-18, $\delta_{\rm C}$ 78.70; C-6', $\delta_{\rm C}$ 71.11) and the sulfate ester (C-5, $\delta_{\rm C}$ 80.33).

Detailed analyses of ¹H–¹H COSY, TOCSY, HSQC, and CH₂-selected E-HSQC^{3,4} data revealed the presence of one isolated methylene (C-55) as well as proton connectivities of the following structural units: (a) from C-1 to C-13, (b) from C-15 to C-27, (d) from C-33 to C-39, (e) from C-41 to C-49, (f) from C-51 to C-53, (g) from C-1' to C-6', and (h) from C-1" to C-6" (Fig. 1). The HMBC spectrum showed the correlations from H₂-13 and H₂-15 to C-14 ($\delta_{\rm C}$ 212.25), indicating that partial structures **a** and **b** were connected through a ketone carbonyl (C-14). The HMBC correlation from H-4 to C-8 revealed an ether linkage between C-4 and C-8 forming a tetrahydropyran ring. ¹H and ¹³C NMR data for C-1–C-18 of colopsinol B (1) including ¹H–¹H coupling constants and NOE data corresponded well to those of C-1–C-18 of colopsinol A.² The epoxide ring at C-22–C-23 was elucidated as *trans* on the basis

of the ¹H–¹H coupling constant ($J_{22,23} = 2.3$ Hz). HMBC correlations observed from H-39 ($\delta_{\rm H}$ 6.12) to C-40 ($\delta_{\rm C}$ 147.26) and from H₂-55 ($\delta_{\rm H}$ 4.92, 2H) to C-39 ($\delta_{\rm C}$ 134.11) and C-41 $(\delta_{\rm C} 37.45)$ suggested the connectivity between partial structures d and e is through the *exo*-methylene unit (C-40–C-55). Of the three disubstituted double bonds, that of C-38-C-39 was elucidated as *E*-geometry by the ${}^{1}\text{H}{-}{}^{1}\text{H}$ coupling constant ($J_{38,39}$ = 15.7 Hz), while two olefins at C-34-C-35 and C-42-C-43 were both assigned to have *E*-geometries from the ¹³C chemical shifts of allylic carbons (C-33 and C-36, $\delta_{\rm C}$ 34.41; C-44, $\delta_{\rm C}$ 39.16). The connection between partial structures e and f through an exo-methylene unit (C-50-C-56) was based on HMBC correlations from H2-56 ($\delta_{\rm H}$ 4.88, 2H) to C-49 ($\delta_{\rm C}$ 43.89), C-50 ($\delta_{\rm C}$ 147.23), and C-51 ($\delta_{\rm C}$ 42.99) and from H₂-51 ($\delta_{\rm H}$ 2.86, 2H) to C-50. The two sugar units g and h were both assigned as β-D-glucose from the ¹H–¹H coupling constants ($J_{1',2'}$ = 7.9 Hz; $J_{1'',2''} = 7.8$ Hz) and ${}^{1}J_{C-H}$ values (C-1' and C-1", 156 Hz) of the anomeric protons as well as chiral HPLC analyses of the 1-methyl-2,3,4,6-O-tetrabenzoyl derivatives of methanolysis products of 1. The ROESY correlation observed from H-6' to H-1" suggested that the sugar part was a β -glucopyranosyl- $(1\rightarrow 6)$ - β -glucopyranoside. This glucoside part was connected to C-18 on the basis of the HMBC correlation observed from H-1' to H-18. The conformation of the tetrahydrofuran ring (C-4–C-8) was assigned as a chair form by the ¹H–¹H coupling constants and NOE data (Fig. 2). The methine signal of C-5 $(\delta_{\rm C} 80.33)$ in 1 was located at lower-field than those (*ca*. $\delta_{\rm C} 70$) of the axially oriented hydroxy-bearing carbons on two tetrahydrofuran rings, suggesting that the sulfate ester was attached to C-5.§ The ROESY correlation from H-45 ($\delta_{\rm H}$ 3.77) to H-48 $(\delta_{\rm H} 4.28)$ implied the presence of a tetrahydrofuran ring with 45,48-syn configuration. Irradiation of H-46 ($\delta_{\rm H}$ 4.06) yielded NOE (5.6%) for H₂-44 ($\delta_{\rm H}$ 2.27), implying the 45,46-*anti* configuration. Although the C₅-linear chain (c) of C-27–C-33 could not be assigned because of heavily overlapped proton and carbon signals, the existence of five methylene carbons for unit c was deduced from FABMS/MS data of the peroxide (3) of colopsinol B (2) as described below. Thus the overall structure of colopsinol B including relative stereochemistry of the tetrahydropyran and tetrahydrofuran rings was concluded to be 1.

HRESIMS data [m/z 1245.6575, (M+Na)⁺, Δ +1.6 mmu] of the colopsinol C (2) revealed the molecular formula to be

[‡] The carbon chemical shift of C-44 ($\delta_{\rm C}$ 39.16) was closer to that of the allylic methylene carbon (C-5; $\delta_{\rm C}$ 36.6) adjacent to the *E*-double bond of 2,2'-bis(tetrahydrofuran) laroxane⁵ than that (C-13; $\delta_{\rm C}$ 29.4) attached to the *Z*-double bond of aplydilactone.⁶

S The axially-oriented hydroxy-bearing positions (C-27; $\delta_{\rm C}$ 69.7 and C-40; 69.5) on two tetrahydropyran rings of luteophanol A.⁷

Table 1 ¹ H and ¹³ C NMR d	lata of colopsinol B (1)
--	--------------------------

Position	$\delta_{ m H}{}^a$	(m, <i>J</i> /Hz)			$\delta_{\rm c}{}^a$	m
1	1.25 ^b	(d, 6.3)			24.81	q
2	3.96	(m)			65.71	d
3	1.89	(m)	1.58	(m)	39.91	t
4	4.55	(dt, 10.8, 3.6)			73.77	d
5	4.36	(brt, 3.2)			80.33	d
6	4.12	(ddd, 10.4, 4.3, 3.2)			67.32	d
7	1.87	(m)	1.78	(m)	33.38	t
8	3.56	(ddd, 9.3, 5.9, 2.2)			74.91	d
9	3.68	(m)			74.66	d
10	1.66	(m)	1.60	(m)	35.24	t
11	1.66	(m)	1.60	(m)	30.55	t
12	4.14	(m)			69.43	d
13	2.67 ^c	(m)			53.29	t
14					212.25	S
15	2.67 ^c	(m)			52.44	t
16	4.47	(m)			66.20	d
17	1.71	(m)	1.63	(m)	43.99	t
18	3.99	(m)			78.70	d
19	1.74	(m)	1.63	(m)	37.56	t
20	1.62	(m)	1.56	(m)	23.55	t
21	1.59°	(m)			33.89	t
22	2.80	(m)			61.31	d
23	2.76	(ddd, 6.8, 5.0, 2.3)			61.62	d
24	1.62	(m)	1.48	(m)	31.54	t
25	1.52	(m)	1.27	(m)	35.14	t
26	1.48	(m)			34.69	d
27	1.37	(m)	1.17	(m)	38.88	t
28	1.33 ^c	(m)			31.86 ^d	t
29	1.33 ^c	(m)			31.66 ^{<i>d</i>}	t
30	1.33°	(m)			31.45 ^d	t
31	1.33°	(m)			30.97 ^d	t
32	1.35°	(m)			28.94	t
33	2.03 ^c	(m)			34.41	t
34	5.46	(m)			132.97	d
35	5.45	(m)			131.52	d
36	2.14 ^c	(m)			34.41	t
37	2.19 ^c	(m)			34.86	t
38	5.78	(dt, 15.7, 6.8)			132.05	d
39	6.12	(d, 15.7)			134.11	d
40					147.26	S
41	2.94 ^c	(d, 6.4)			37.45	t
42	5.60	(m)			132.71	d
43	5.55	(m)			129.28	d
44	2.27 °	(m)			39.16	t
45	3.77	(dt, 2.9, 6.1)			88.46	d
46	4.06	(ddd, 6.4, 2.9, 2.5)			76.75	d
47	1.92	(ddd, 13.1, 5.5, 2.5)	1.73	(ddd, 13.1, 9.8, 6.4)	42.39	t
48	4.28	(m)			79.03	d
49	2.38	(dd, 14.2, 6.7)	2.24	(m)	43.89	t
50					147.23	S
51	2.86 ^c	(d, 7.0)			42.99	t
52	5.86	(ddt, 17.1, 10.1, 6.9)			138.35	d
53	5.09 ^c	(m)			117.55	t
54	0.93 ^b	(d, 6.5)			20.84	q
55	4.92 ^c	(m)			115.35	t
56	4.88 ^c	(m)			113.72	t
1'	4.42	(d, 7.9)			105.03	d
2'	3.23	(m)			76.27	d
3'	3.40	(m)			78.81	d
4'	3.56	(m)			72.49	d
5'	3.51	(m)			77.84	d
6'	4.17	(dd, 11.7, 2.1)	3.82	(dd, 11.7, 5.9)	71.11	t
1″	4.47	(d, 7.8)			105.86	d
2″	3.23	(m)			75.97	d
3″	3.40	(m)			78.81	d
4″	3.56	(m)			72.49	d
5″	3.31	(m)			78.90	d
6″	3.91	(dd, 11.9, 2.1)	3.71	(dd, 11.9, 5.3)	63.63	t
^{<i>a</i>} In CD ₃ OD. ^{<i>b</i>} 3H. ^{<i>c</i>}	2H. ^d These signa	als may be interchangeable.				

 $C_{62}H_{103}O_{20}SNa$. ¹H and ¹³C NMR data of **2** containing a ketone, two sp² quaternary carbons, seven sp² methines, three sp² methylenes, nineteen oxymethines, one oxymethylene, one sp³ methines, twenty-six sp³ methylenes, and two methyl groups were similar to those of **1**, except for a lack of resonances for

one glucose. Detailed analyses of 2D NMR data including ¹H–¹H COSY, TOCSY, HSQC, and CH₂-selected E-HSQC^{3,4} spectra revealed that the C-1–C-53 part of **2** possessed the same overall structure as that of **1**. The C-6' (δ_C 63.71) of **2** resonated in higher-field than C-6' (δ_C 71.11) of **1**. Thus the structure of

colopsinol C (2) was elucidated to be the monodeglucosyl form of colopsinol B (1).

The structure of the peroxide (3) of colopsinol C (2) was deduced from detailed analyses of 2D NMR data (${}^{1}H{-}^{1}H$ COSY, TOCSY, HSQC, CH₂-selected E-HSQC,^{3,4} and HMBC spectra) and the FABMS/MS spectrum (Fig. 3). The presence of the six-memberd peroxide ring was supported by the ion peaks at *m*/*z* 995 and 955.

Colopsinols B (1) and C (2) are new polyhydroxy compounds possessing a C₅₃-linear carbon chain including three C₁ branches as well as a tetrahydropyran, a tetrahydrofuran, and an epoxide ring, six hydroxy groups, a glucoside moiety, and a sulfate ester. Colopsinol C (2) exhibited cytotoxicity against L1210 murine leukemia cells *in vitro* with the IC₅₀ value of 7.8 μ g mL⁻¹, while colopsinol B (1) did not show such cytotoxicity (IC₅₀ > 10 μ g mL⁻¹).

Experimental

Cultivation and isolation

The dinoflagellate *Amphidinium* sp. (strain number Y-5) was unialgally cultured at 25 °C for two weeks in seawater medium enriched with 1% ES supplement. The harvested cells of the cultured dinoflagellate (1205 g, wet weight, from 4960 L of culture) were extracted with MeOH–toluene (3:1, 3 L × 3). After addition of 1 M NaCl (1 L), the mixture was extracted



---> NOE ---> ROESY

Fig. 2 Relative stereochemistry of tetrahydropyran (A) and tetrahydrofuran (B) rings in colopsinol B (1). The coupling for this moiety (H–H in Hz) are as follows: 4-5=3.6, 5-6=3.2, $6-7\alpha=10.4$, $6-7\beta=4.3$, $7\alpha-8=9.3$, $7\beta-8=2.2$, 45-46=2.9, $46-47\alpha=6.4$ Hz, $46-47\beta=2.5$ Hz, $47\alpha-48=9.8$ Hz, and $47\beta-48=5.5$ Hz.

with toluene (4 L × 3). The toluene-soluble fraction was evaporated under reduced pressure to give a residue (44.4 g), which was partially (26.7 g) subjected to a silica gel column eluted with CHCl₃-MeOH (95:5 \rightarrow 0:100). Part (2.33 g) of the fraction (3.3 g) eluted with MeOH was purified by gel filtration on Sephadex LH-20 (CHCl₃-MeOH, 1:1), centrifugal partition chromatography (ascended mode, CHCl₃-MeOH-H₂O, 5:6:4), and C₁₈ column chromatography (*i*-PrOH-H₂O, 40:60) to afford colopsinols B (1, 5.0 mg, 0.001%, wet weight) and C (2, 1.0 mg, 0.0002%), and the peroxide (3, 4.1 mg, 0.0008%) of colopsinol C (2).

Colopsinol B (1). A colorless amorphous solid; λ_{max}/nm (MeOH) 232 ($\epsilon/dm^3 mol^{-1} cm^{-1} 13000$); ν_{max}/cm^{-1} (KBr) 3430, 1700 and 1250; ¹H and ¹³C NMR (Table 1); m/z (ESIMS) 1362 (M - Na)⁻; m/z (FABMS) 80 (SO₃⁻), 97 (HSO₄⁻) and 1362 (M - Na)⁻; m/z (HRESIMS) 1407.7040 [(M + Na)⁺. Calc. for C₆₈H₁₁₃O₂₅SNa₂;1407.7087].

Colopsinol C (2). A colorless amorphous solid; UV (MeOH) $\lambda_{\rm max}/{\rm nm}$ 232 ($\epsilon/{\rm dm}^3$ mol⁻¹ cm⁻¹ 10000); IR (KBr) $v_{\rm max}/{\rm cm}^{-1}$ 3430, 1700, and 1250; $\delta_{\rm H}$ (MeOH- d_4) 0.93 (3H, d, J = 6.3 Hz, H₃-54), 1.17 (1H, m, H-27), 1.25 (3H, d, *J* = 6.3 Hz, H₃-1), 1.29 (1H, m, H-25), 1.33 (8H, m, H₂-28, H₂-29, H₂-30 and H₂-31), 1.34 (2H, m, H₂-32), 1.38 (1H, m, H-27), 1.48 (1H, m, H-26), 1.49 (1H, m, H-24), 1.55 (2H, m, H-20 and H-25), 1.57 (1H, m, H-21), 1.59 (1H, m, H-3), 1.62 (4H, m, H-10, H-11, H-20 and H-21), 1.63 (1H, m, H-17), 1.64 (1H, m, H-24), 1.65 (1H, m, H-19), 1.66 (1H, m, H-11), 1.67 (1H, m, H-10), 1.71 (1H, m, H-17), 1.72 (1H, m, H-47), 1.77 (1H, m, H-19), 1.78 (1H, m, H-7), 1.87 (1H, m, H-7), 1.89 (1H, m, H-3), 1.92 (1H, m, H-47), 2.03 (2H, m, H₂-33), 2.12 (2H, m, H₂-36), 2.20 (2H, m, H₂-37), 2.23 (1H, m, H-49), 2.27 (2H, m, H₂-44), 2.38 (1H, dd, J = 14.0 and 6.5 Hz, H-49), 2.67 (4H, m, H2-13 and H2-15), 2.74 (1H, m, H-23), 2.78 (1H, m, H-22), 2.86 (2H, br d, J = 6.7 Hz, H₂-51), 2.93 (2H, br d, J = 6.0 Hz, H₂-41), 3.23 (1H, m, H-2'), 3.31 (1H, m, H-5'), 3.35 (1H, m, H-4'), 3.40 (1H, m, H-3'), 3.56 (1H, ddd, J = 9.0, 5.6 and 2.4 Hz, H-8), 3.69 (1H, m, H-9), 3.71 (1H, dd, J = 11.8 and 5.0 Hz, H-6'), 3.78 (1H, dt, J = 3.1 and 6.0 Hz, H-45), 3.89 (1H, br d, *J* = 11.8 Hz, H-6'), 3.98 (1H, m, H-18), 3.99 (1H, m, H-2), 4.06 (1H, dt, J = 6.4 and 3.0 Hz, H-46), 4.12 (1H, dt, J = 10.0 and 3.8 Hz, H-6), 4.13 (1H, m, H-12), 4.29 (1H, m, H-48), 4.36 (1H, br t, J = 3.0 Hz, H-5), 4.46 (1H, d, H)J = 8.0 Hz, H-1'), 4.47 (1H, m, H-16), 4.53 (1H, dt, J = 10.6 and



Fig. 3 Fragmentation patterns observed in negative ion FABMS/MS spectrum of the peroxide (3) of colopsinol C (2) (precursor ion, m/z 1231).

3.7 Hz, H-4), 4.87 (2H, m, H₂-56), 4.92 (2H, m, H₂-55), 5.09 (2H, m, H₂-53), 5.44 (1H, m, H-35), 5.46 (1H, m, H-34), 5.56 (1H, m, H-43), 5.60 (1H, m, H-42), 5.78 (1H, dt, J = 15.7 and 6.8 Hz, H-38), 5.86 (1H, m, H-52) and 6.12 (1H, d, J = 15.7 Hz, H-39); δ_c(MeOH-d₄) 20.81 (q, C-54), 23.46 (t, C-20), 24.80 (q, C-1), 28.91 (t, C-32), 30.52 (t, C-11), 30.94 (t, C-31), 31.43 (t, C-30), 31.52 (t, C-24), 31.59 (t, C-29), 31.83 (t, C-28), 33.37 (t, C-7), 33.83 (t, C-21), 34.39 (t, C-36), 34.41 (t, C-33), 34.64 (d, C-26), 34.84 (t, C-37), 35.07 (t, C-25), 35.24 (t, C-10), 37.45 (t, C-41), 37.52 (t, C-19), 38.84 (t, C-27), 39.15 (t, C-44), 39.94 (t, C-3), 42.38 (t, C-47), 42.99 (t, C-51), 43.88 (t, C-49), 44.01 (t, C-17), 52.43 (t, C-15), 53.30 (t, C-13), 61.37 (d, C-22), 61.46 (d, C-23), 63.71 (t, C-6'), 65.72 (d, C-2), 66.16 (d, C-16), 67.35 (d, C-6), 69.44 (d, C-12), 72.51 (d, C-4'), 73.75 (d, C-4), 74.65 (d, C-9), 74.93 (d, C-8), 76.34 (d, C-2'), 76.75 (d, C-46), 78.65 (d, C-18), 79.00 (d, C-5'), 79.03 (d, C-3'), 79.08 (d, C-48), 88.46 (d, C-45), 105.41 (d, C-1'), 113.72 (t, C-56), 115.35 (t, C-55), 117.54 (t, C-53), 129.27 (d, C-43), 131.52 (d, C-35), 132.05 (d, C-38), 132.97 (d, C-34), 134.10 (d, C-39), 137.72 (d, C-42), 138.35 (d, C-52), 147.23 (s, C-40), 147.23 (s, C-50) and 212.28 (s, C-14); m/z (ESIMS) 1199 (M – Na)⁻; m/z (HRESIMS) 1245.6575 [$(M + Na)^+$. Calc. for C₆₂H₁₀₃O₂₀SNa₂: 1245.6559].

The peroxide (3) of colopsinol C (2). A colorless amorphous solid; $v_{\text{max}}/\text{cm}^{-1}$ (KBr) 3430, 1700 and 1250; δ_{H} (MeOH- d_{4}) 0.93 $(3H, d, J = 6.3 Hz, H_3-54), 1.17 (1H, m, H-27), 1.25 (3H, J-27), 1.25 ($ d, J = 6.3 Hz, H₃-1), 1.27 (1H, m, H-25), 1.33 (8H, m, H₂-28, H_2 -29, H_2 -30 and H_2 -31), 1.34 (2H, m, H_2 -32), 1.36 (1H, m, H-27), 1.48 (1H, m, H-26), 1.49 (1H, m, H-24), 1.52 (1H, m, H-25), 1.55 (1H, m, H-20), 1.58 (1H, m, H-3), 1.60 (4H, m, H-11, H₂-21 and H-37), 1.61 (2H, m, H-10 and H-24), 1.62 (2H, m, H-17 and H-20), 1.65 (1H, m, H-19), 1.66 (3H, m, H-10, H-11 and H-37), 1.71 (1H, m, H-17), 1.75 (1H, m, H-47), 1.76 (1H, m, H-19), 1.77 (1H, m, H-7), 1.87 (1H, m, H-7), 1.89 (1H, m, H-3), 1.94 (1H, m, H-47), 2.03 (2H, m, H₂-33), 2.12 (2H, m, H₂-36), 2.23 (1H, m, H-49), 2.28 (2H, m, H₂-44), 2.38 (1H, dd, J = 14.0 and 6.5 Hz, H-49), 2.67 (4H, m, H₂-13 and H₂-15), 2.75 (1H, ddd, J = 6.2, 5.0 and 1.9 Hz, H-23), 2.78 (1H, m, H-22), 2.79 (2H, br d, *J* = 4.7 Hz, H₂-41), 2.86 (2H, br d, *J* = 6.7 Hz, H₂-51), 3.22 (1H, m, H-2'), 3.31 (1H, m, H-4'), 3.41 (1H, m, H-3'), 3.47 (1H, m, H-5'), 3.55 (1H, ddd, J = 8.7, 5.6 and 3.1 Hz, H-8), 3.70 (1H, m, H-9), 3.76 (1H, dt, J = 3.1 and 6.0 Hz, H-45), 3.71 (1H, dd, J = 11.8 and 5.0 Hz, H-6'), 3.89 (1H, dd, J = 11.8 and 2.5 Hz, H-6'), 3.98 (2H, m, H-2 and H-18), 4.05 (1H, dt, J = 6.4 and 3.0 Hz, H-46), 4.11 (1H, dt, J = 10.0 and 3.8 Hz, H-6), 4.13 (1H, m, H-12), 4.28 (1H, ddd, J = 12.3, 9.8 and 6.2 Hz, H-48), 4.35 (1H, br t, J = 3.1 Hz, H-5), 4.38 (1H, m, H-55), 4.44 (1H, m, H-55), 4.45 (1H, d, J = 8.0 Hz, H-1'), 4.47 (1H, m, H-16), 4.52 (1H, dt, J = 10.6 and 3.7 Hz, H-4), 4.52 $(1H, m, H-38), 4.87 (2H, br d, J = 4.7 Hz, H_2-56), 5.09 (2H, m,$ H₂-53), 5.46 (1H, m, H-34), 5.47 (1H, m, H-35), 5.57 (2H, m, H-42 and H-43), 5.69 (1H, br s, H-39) and 5.86 (1H, m, H-52); $\delta_{\rm C}$ (MeOH- d_4) 20.80 (q, C-54), 23.44 (t, C-20), 24.81 (q, C-1), 28.84 (t, C-32), 30.14 (t, C-36), 30.52 (t, C-11), 31.36 (t, C-28), 31.42 (t, C-29), 31.51 (t, C-30), 31.51 (t, C-31), 31.55 (t, C-24), 33.33 (t, C-7), 33.81 (t, C-21), 34.36 (t, C-33), 34.59 (d, C-26), 34.85 (t, C-37), 35.03 (t, C-25), 35.21 (t, C-10), 37.35 (t, C-41), 37.49 (t, C-19), 38.78 (t, C-27), 39.03 (t, C-44), 39.84 (t, C-3), 42.41 (t, C-47), 42.94 (t, C-51), 43.89 (t, C-17), 43.89 (t, C-49), 52.46 (t, C-15), 53.31 (t, C-13), 61.32 (d, C-22), 61.40 (d, C-23), 63.69 (t, C-6'), 65.67 (d, C-2), 66.14 (d, C-16), 67.26 (d, C-6), 69.41 (d, C-12), 72.48 (d, C-4'), 73.63 (t, C-55), 73.89 (d, C-4), 74.71 (d, C-9), 74.79 (d, C-8), 76.31 (d, C-2'), 76.80 (d, C-46), 78.61 (d, C-18), 78.80 (d, C-5'), 78.96 (d, C-3'), 78.96 (d, C-48), 79.75 (d, C-38), 80.32 (d, C-5), 88.25 (d, C-45), 105.35 (d, C-1'), 113.72 (t, C-56), 117.55 (t, C-53), 124.00 (s, C-40), 130.67 (d, C-43), 131.00 (d, C-42), 131.31 (d, C-35), 133.16 (d, C-34), 136.93 (d, C-39), 138.32 (d, C-52), 147.21 (s, C-50) and 212.24 (s, C-14); *m*/*z* (ESIMS) 1231 (M – Na)[–]; *m*/*z* (HRESIMS) 1277.6844 [(M + Na)⁺. Calc. for $C_{62}H_{103}O_{22}SNa_2$: 1277.6820].

Determination of stereochemistry of the sugar units of colopsinols B (1) and C (2) by chiral HPLC

Colopsinol B (1, 0.3 mg) or C (2, 0.3 mg) was treated with 3% HCl-MeOH (300 µL) at 110 °C for 1 h. After the solvent was removed by a nitrogen stream, to the residue was added CHCl₃ (100 µL) and the CHCl₃ solution was extracted with H_2O (100 $\mu L \times 3$). The aqueous fraction evaporated *in vacuo* was treated with pyridine (100 μ L), triethylamine (15 μ L), and benzovl chloride (15 μ L) at room temperature for 21 h. After addition of MeOH (100 µL), the reaction mixture was extracted with hexane (100 μ L × 3). The hexane-soluble fraction was evaporated in vacuo to afford the 1-methyl-2,3,4,6-O-tetrabenzoyl derivative of each sugar unit. Authentic D- and L-glucose were treated with HCl-MeOH and then benzoyl chloride as described above to give the 1-methyl-2,3,4,6-Otetrabenzoyl derivatives of D- and L-glucose, respectively. The 1-methyl-2,3,4,6-O-tetrabenzoyl derivatives were subjected to chiral HPLC analyses using Chiralpak OP(+) (Daicel Chemical Industry Ltd., 4.6 × 250 mm; MeOH; flow rate, 0.5 mL min⁻¹; UV detection at 254 nm). The retention times of the 1-methyl-2,3,4,6-O-tetrabenzoyl derivative of the methanolysis product of **1** or **2** were found to be 23.8 min, while the retention times of the 1-methyl-2,3,4,6-O-tetrabenzoyl derivatives of authentic D- and L-glucose were found to be 23.8 and 25.8 min, respectively.

Acknowledgements

We thank Drs J. Kawabata and E. Fukushi, Faculty of Agriculture, Hokkaido University, for use of E-HSQC pulse program, and Mr K. Watanabe, GC-MS & NMR laboratory, Faculty of Agriculture, Hokkaido University, for measurements of ESIMS. This work was partly supported by a Grantin-Aid for Scientific Research from the Ministry of Education, Science, Sports, and Culture of Japan.

References

- 1 M. Ishibashi and J. Kobayashi, *Heterocycles*, 1997, 44, 543 and references cited therein.
- 2 J. Kobayashi, T. Kubota, M. Takahashi, M. Ishibashi, M. Tsuda and H. Naoki, J. Org. Chem., 1999, 64, 1478.
- 3 D. G. Davis, J. Magn. Reson., 1991, 91, 665.
- 4 E. Fukushi, S. Tanabe, M. Watanabe and J. Kawabata, *Magn. Reson. Chem.*, 1998, **36**, 741.
- 5 I. K. Kim, M. R. Brennan and K. L. Erickson, *Tetrahedron Lett.*, 1989, **30**, 1757.
- 6 M. Ojika, Y. Yoshida, Y. Nakayama and K. Yamada, *Tetrahedron Lett.*, 1990, **31**, 4907.
- 7 Y. Doi, M. Ishibashi, H. Nakamichi, T. Kosaka, T. Ishikawa and J. Kobayashi, J. Org. Chem., 1997, **62**, 3820.

Paper 9/06296C